



(12) **EUROPEAN PATENT APPLICATION**

(43) Date of publication:
11.01.2006 Bulletin 2006/02

(51) Int Cl.:
C07D 211/22 (2006.01) A61K 31/4409 (2006.01)
A61P 11/08 (2006.01)

(21) Application number: **05011652.4**

(22) Date of filing: **31.05.2005**

(84) Designated Contracting States:
AT BE BG CH CY CZ DE DK EE ES FI FR GB GR
HU IE IS IT LI LT LU MC NL PL PT RO SE SI SK TR
Designated Extension States:
AL BA HR LV MK YU

- Barreca, Giuseppe
23874 Montecatini (LC) (IT)
- Allegrini, Pietro
20097 San Donato Milanese (MI) (IT)
- Ventimiglia, Gianpiero
20092 Cinisello Balsamo (MI) (IT)

(30) Priority: **08.06.2004 IT mi20041143**

(71) Applicant: **Dipharma S.p.A.**
33036 Mereto di Tomba (Udine) (IT)

(74) Representative: **Minoja, Fabrizio**
Blanchetti Bracco Minoja S.r.l.
Via Plinio, 63
20129 Milano (IT)

(72) Inventors:
• **Castaldi, Graziano**
28072 Briona (NO) (IT)

(54) **Fexofenadine polymorphs and process for the preparation thereof**

(57) The invention provides novel crystalline forms of fexofenadine hydrochloride, a process for the preparation of the novel forms and of the known form A, and their use in therapy.

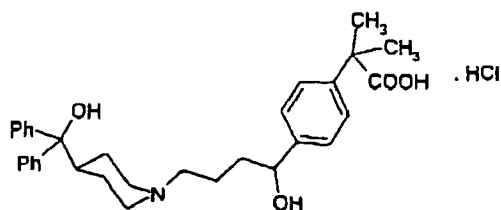
Description

FIELD OF THE INVENTION

[0001] The present invention relates to novel crystalline forms of fexofenadine hydrochloride, in particular a hydrated form and a solvate with acetonitrile, their use in therapy and a process for their preparation. The invention further relates to a process for the preparation of the known polymorphic form A of anhydrous fexofenadine hydrochloride.

TECHNOLOGICAL BACKGROUND

[0002] Fexofenadine hydrochloride (4-[4-[4-(hydroxydiphenylmethyl)-1-piperidinyl]-1-hydroxybutyl]- α,α -dimethylbenzeneacetic acid hydrochloride)



is an antihistaminic, antiallergic and bronchodilator medicament, marketed in the USA under the trade-name Allegra®.

[0003] WO 02/066429 discloses anhydrous fexofenadine hydrochloride polymorphic form A and a process for the preparation thereof.

[0004] US 2003/0021849 discloses other crystalline forms of fexofenadine hydrochloride, such as solvates with methyl butyl ether (MTBE) and cyclohexane, referred to as MBTE form IX and cyclohexane form IX, as well as solvates with ethyl acetate, referred to as forms XIV and XV.

[0005] WO 95/31437 discloses the preparation of polymorphs, or pseudopolymorphs and hydrates of fexofenadine hydrochloride (form II and IV) and their conversion to anhydrous polymorphic forms (form I and II) by azeotropic distillation or recrystallization to "minimize the water content".

[0006] Different forms of biologically active compounds, such as polymorphs, are known to have different bioavailability, release time and solubility, which allow, for example, dose reduction or prolonged administration intervals. Moreover, the different physical properties that are often associated to different physical forms of medicaments can be advantageously exploited in the manufacture of pharmaceutical formulations.

[0007] There is therefore the need to provide novel polymorphic forms of biologically active compounds with advantageous properties.

SUMMARY OF THE INVENTION

[0008] In a first aspect the invention provides a novel approximately monohydrate crystalline form of fexofenadine hydrochloride, hereinafter referred to as form B, and a method for its preparation.

[0009] In a second aspect the invention provides a novel crystalline form of fexofenadine hydrochloride, approximately monosolvate with acetonitrile, hereinafter referred to as form C, and a method for its preparation.

[0010] In a further aspect, the invention provides the use of fexofenadine hydrochloride form C for the preparation of the known form A.

[0011] Furthermore, the invention provides a pharmaceutical composition containing fexofenadine hydrochloride form B and/or C as active ingredient in admixture with an excipient and/or carrier, and its use in therapy.

DESCRIPTION OF THE FIGURES

[0012] The novel crystalline forms B and C and the known form A were characterized by XRPD (X-ray powder diffraction) technique. The X-ray diffraction spectra (XRPD) were registered with an APD 2000 θ/θ automatic diffractometer for powders and liquids (Ital-Structures), under the following operative conditions: radiation $\text{CuK}\alpha$ ($\lambda = 1,5418 \text{ \AA}$), scanning with a 0.03°C angular step for 1 sec. The DSC (Differential Scanning Calorimetry) spectra were recorded with a DSC 822e calorimeter (Mettler-Toledo), in a temperature range of $30\text{--}250^\circ\text{C}$, with a scanning rate of $10^\circ\text{C}/\text{minute}$. The $^1\text{H-NMR}$

spectra were recorded with a Varian Mercury 300 spectrometer, using DMSO-d₆ as the solvent.

Figure 1. XRPD spectrum (X-ray powder diffraction) of fexofenadine hydrochloride form B.

Figure 2. DSC spectrum of fexofenadine hydrochloride form B.

Figure 3. XRPD spectrum of fexofenadine hydrochloride form C.

Figure 4. ¹H-NMR spectrum of fexofenadine hydrochloride form C.

DETAILED DISCLOSURE OF THE INVENTION

[0013] According to the present invention, "approximately monohydrate" means that the crystalline solid has a water content of about 0.8-1.2 mols per mol of fexofenadine hydrochloride, preferably from about 0.9 to 1.1 mols. "Approximately monosolvate" means that the crystalline solid has an acetonitrile content of about 0.8-1.2 mols of solvent per mol fexofenadine hydrochloride, preferably from 0.9 to 1.1 mols.

[0014] As used in the following, the term precipitation (or "precipitate") has the same meaning as crystallization (or "crystal"), and indicates the obtainment of a solid form.

[0015] The present invention relates to a novel approximately monohydrate, crystalline form of fexofenadine hydrochloride, hereinafter defined as form B, having an XRPD spectrum substantially as illustrated in Figure 1, wherein the more intense diffraction peaks fall at 23.6; 10.1; 7.9; 5.2; 4.7 and 4.4 in 2θ, and a DSC spectrum having three endothermic peaks at about 80.27; 109.27 and 149.14°C; said form B being prepared through a process comprising the following steps:

- preparation of a solution of fexofenadine hydrochloride in a methanol/water mixture;
- cooling of the mixture at a temperature lower than -5°C; and
- separation of the precipitate.

[0016] Fexofenadine hydrochloride can be obtained by reduction of 4-[4-(hydroxydiphenylmethyl)-1-piperidinyl]-1-oxobutyl]-α,α-dimethylbenzeneacetic acid, or an alkyl ester thereof, with a suitable reducing agent, optionally followed by hydrolysis, as known in the art.

[0017] The methanol/water ratio in the mixture containing the solution of fexofenadine hydrochloride is for example between 0.8 and 1.25 v/v, preferably from 0.9 and 1.1 v/v, in particular around 1 v/v.

[0018] The mixture is usually cooled slowly, preferably at a temperature ranging from -10 to -25°C, more preferably from -12 to -20°C.

[0019] The precipitate is filtered off or decanted, preferably filtered off and, after washing with a water-methanol mixture in a volume ratio of 0.8 to 1.25 v/v, preferably from 0.9 to 1.1 v/v, in particular around 1 v/v, is dried to constant weight at 50-60°C, preferably under vacuum at approximately 40°C. Karl-Fischer analysis of the resulting product shows that novel crystalline form B of fexofenadine hydrochloride hydrate has a water content ranging from about 3 to 5%, so that it can be properly defined as "approximately monohydrate".

[0020] In a second aspect, the invention provides a novel crystalline form of fexofenadine hydrochloride, approximately monosolvate with acetonitrile, hereinafter referred to as form C, having an XRPD spectrum substantially as illustrated in figure 3, wherein the most intense diffraction peaks fall at 7.0; 11.6; 15.4; 17.3; 18.0 and 20.5 in 2θ; and a ¹H-NMR spectrum substantially as illustrated in figure 4. The novel form can be prepared by means of a process comprising the following steps:

- preparation of a dispersion of fexofenadine hydrochloride hydrate in acetonitrile;
- heating of the dispersion at a temperature ranging from about 70°C to the reflux temperature;
- cooling of the dispersion at a temperature lower than -5°C; and
- separation of the precipitate.

[0021] Fexofenadine hydrochloride hydrate, used as starting material, can be prepared according to known methods, for example as disclosed in WO 95/31437, or can be the novel form B which is disclosed herein.

[0022] The acetonitrile dispersion is heated for a time ranging from about 15 minutes to one hour, preferably from 30 to 45 minutes, then cooled preferably to a temperature ranging from about -15 to -10°C, for a time sufficient to obtain complete precipitation of fexofenadine hydrochloride, approximately monosolvate with acetonitrile, typically from about 15 minutes to 8 hours, preferably for about 5 hours.

[0023] According to a preferred aspect of the invention, the dispersion of fexofenadine hydrochloride hydrate in acetonitrile is obtained pouring fexofenadine hydrochloride hydrate in acetonitrile, previously heated to about 35-60°C, preferably to about 40-55°C, in particular 45°C, in a weight/volume ratio ranging from about 1/8 to 1/15, preferably of about 1/10. Fexofenadine hydrochloride hydrate is added to acetonitrile until all the fexofenadine is dissolved, keeping the temperature between about 70°C and the reflux temperature. During this phase fexofenadine hydrochloride form C

starts to crystallise and precipitate. The acetonitrile dispersion is cooled to a temperature lower than -5°C, preferably to about -15/-10°C, in about four hours. The precipitate is then separated with known techniques, for example filtration or decantation of the solvent, preferably filtration. The product is then dried at a temperature of about 60°C or lower, preferably under vacuum at approximately 50°C for 10 hours. NMR analysis of the resulting product shows that novel crystalline form C of fexofenadine hydrochloride has an acetonitrile content ranging from about 6.5 to 7.5%, therefore it can be properly defined as "approximately monosolvate".

[0024] The skilled chemist will note that the conditions to obtain the dissolution of fexofenadine hydrochloride in methanol/water, or the dispersion of fexofenadine monohydrate in acetonitrile, can be modified without altering the form of the resulting polymorph. Likewise, the crystallization techniques may slightly differ from those herein described without altering the polymorphic form obtained. For example, crystallization can be promoted by addition of pre-formed crystals.

[0025] A further object of the invention is a process for the preparation of anhydrous fexofenadine hydrochloride form A, disclosed in WO 02/066429, comprising removing acetonitrile from crystals of fexofenadine hydrochloride form C, as herein defined.

[0026] The removal of acetonitrile from fexofenadine hydrochloride form C can be carried out with any suitable physical means, preferably by heating under vacuum at about 80-110°C for a time ranging from about 18 to 70 hours, more preferably at about 100°C under vacuum for about 24-30 hours.

[0027] The novel fexofenadine hydrochloride forms B and C are useful for the administration of fexofenadine in mammals in the need of said treatment. They can be administered alone or in combination or in admixture with one or more known polymorphic forms of fexofenadine hydrochloride, for example those disclosed in WO 02/066429, US 2003/0021849 or WO 95/31437. The content of each form in the mixtures depends on their physical and biological properties and will be determined by the skilled person. A variety of pharmaceutical compositions can be prepared for the administration in humans or animals, according to known techniques. The amount of fexofenadine hydrochloride in capsules, tablets, sugar-coated pills or other forms for the single administration ranges from about 30 to about 180 mg per dose unit. Therefore, the invention also provides a pharmaceutical composition comprising, as active ingredient, fexofenadine hydrochloride form B and/or C, or mixtures of at least one of them with one or more known polymorphic forms of fexofenadine hydrochloride in admixture with a suitable carrier and/or excipient. Said pharmaceutical composition can also optionally contain a therapeutically effective amount of pseudoephedrine.

[0028] The following examples illustrate the invention.

Example 1: Preparation of fexofenadine hydrochloride form B

[0029] 278 g of 4-14-[4-(hydroxydiphenylmethyl)-1-piperidinyl]-1-oxobutyl]- α,α -dimethylbenzeneacetic acid, 355 ml of water, 765 ml of methanol and 22.6 g of sodium hydroxide pellets are loaded in a 3 litres four-neck flask. The resulting mixture is heated to 40°C, then a solution of sodium borohydride (9.8 g) in 47 ml of water is added. When the reduction reaction is complete, the mixture is added with 40 ml of acetone and the pH is adjusted to a value ranging from 2.5 to 3 with 36% w/w hydrochloric acid. Thereafter, 140 ml of methanol and 450 ml of water are added, keeping the temperature at 40°C. The resulting mixture is slowly cooled to -15°C and the resulting precipitate is filtered and washed with a 1/1 water/methanol mixture. The resulting product is dried under vacuum at 40°C to constant weight. 265 g of fexofenadine hydrochloride are obtained with a 4.1 % Karl-Fischer water content (therefore it can be referred to as monohydrate) and substantially characterised by the XRPD and DSC spectra reported in Figures 1 and 2.

Example 2: Preparation of fexofenadine hydrochloride form C

[0030] 2650 ml of acetonitrile are loaded in a four-necked 3 litres flask, then heated to 45°C. 265 g of fexofenadine hydrochloride hydrate is added under stirring and the suspension is refluxed at (80-82°C) for about one hour. The suspension is cooled to about -15/-10°C in about 4 hours. The precipitated solid is filtered and washed with acetonitrile (2 x 80 ml) to obtain 285 g of wet product which, after drying under vacuum at 50°C for 10 hours, yields 262 g of fexofenadine hydrochloride monosolvate with acetonitrile, having a XRPD spectrum and a ¹H-NMR spectrum substantially as illustrated in Figures 3 and 4 respectively and a 0.25% Karl-Fischer water content.

Example 3: Preparation of fexofenadine hydrochloride form A

[0031] 262 g of fexofenadine hydrochloride form C, obtained as described in Example 2, are ground and dried under vacuum at 100°C for 24 hours, 244 g of fexofenadine hydrochloride anhydrous form A are obtained, having a melting point between 153 and 156°C and a XRPD spectrum having substantially the same characteristics as those reported in table 1 of WO 02/066429,

Claims

1. Approximately monohydrate crystalline form of fexofenadine hydrochloride, having an XRPD spectrum substantially as reported in Figure 1.
2. The crystalline form of fexofenadine hydrochloride according to claim 1, having an XRPD spectrum wherein the most intense diffraction peaks fall at 23.6; 10.1; 7.9; 5.2; 4.7 and 4.4 in 2θ .
3. The crystalline form of fexofenadine hydrochloride according to claim 1 or 2, having a DSC spectrum characterized by three endothermic peaks at about 80.27; 109.27 and 149.14°C.
4. A process for the preparation of approximately monohydrate fexofenadine hydrochloride as defined in claim 1, comprising the following steps:
 - preparation of a solution of fexofenadine hydrochloride in a methanol/water mixture;
 - cooling of the mixture at a temperature lower than -5°C; and
 - separation of the precipitate.
5. The process according to claim 4, wherein the volume ratio in the methanol/water mixture is between 0.8 and 1.25 v/v.
6. The process according to claim 4 or 5, wherein the mixture is cooled to a temperature ranging from -10 to -25°C.
7. Crystalline form of fexofenadine hydrochloride, approximately monosolvate with acetonitrile, having an XRPD spectrum substantially as that reported in figure 3.
8. The crystalline form of fexofenadine hydrochloride according to claim 7, having an XRPD spectrum wherein the most intense diffraction peaks fall at 7.0; 11.6; 15.4; 17.3; 18.0 and 20.5 in 2θ .
9. The crystalline form of fexofenadine hydrochloride according to claim 7 or 8, having a $^1\text{H-NMR}$ spectrum substantially as that reported in Figure 4.
10. A process for the preparation of fexofenadine hydrochloride, approximately monosolvate with acetonitrile, as defined in claim 7, comprising the following steps:
 - preparation of a dispersion of fexofenadine hydrochloride hydrate in acetonitrile;
 - heating of the acetonitrile dispersion at a temperature ranging from about 70°C to the reflux temperature;
 - cooling of the acetonitrile dispersion to a temperature lower than -5°C; and
 - separation of the precipitate.
11. The process according to claim 10, wherein the dispersion of fexofenadine hydrochloride hydrate in acetonitrile is obtained pouring fexofenadine hydrochloride hydrate in acetonitrile, previously heated to about 35-60°C, in a weight/volume ratio ranging from about 1/8 to 1/15.
12. The process according to claim 10 or 11, wherein the acetonitrile dispersion is cooled to a temperature of about -15/-10°C.
13. A process for the preparation of anhydrous fexofenadine hydrochloride form A, comprising the removal of acetonitrile from crystals of approximately monosolvate fexofenadine hydrochloride with acetonitrile, as defined in claim 7.
14. The process according to claim 13, wherein acetonitrile is removed by heating under vacuum at about 80-110°C for a time ranging from about 18 to 70 hours.
15. Pharmaceutical composition comprising, as active ingredient, approximately monohydrate fexofenadine hydrochloride, as defined in claim 1, and/or fexofenadine hydrochloride, approximately monosolvate with acetonitrile, as defined in claim 7, or mixtures of at least one of them with one or more known polymorphic forms of fexofenadine hydrochloride in admixture with a carrier and/or excipient.
16. The pharmaceutical composition according to claim 15, further containing pseudoephedrine, as active ingredient.

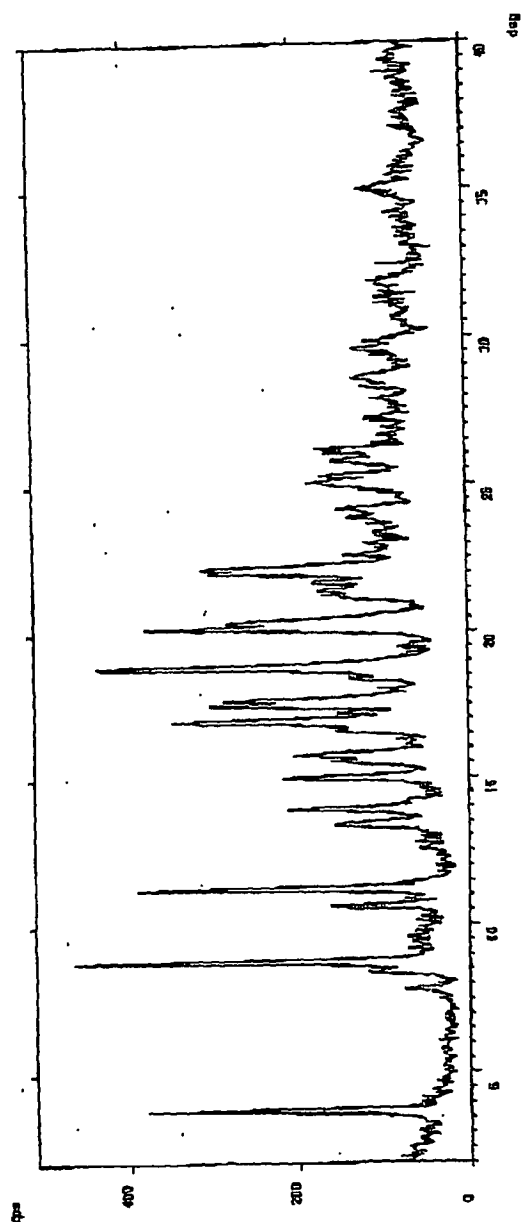


FIGURE 1

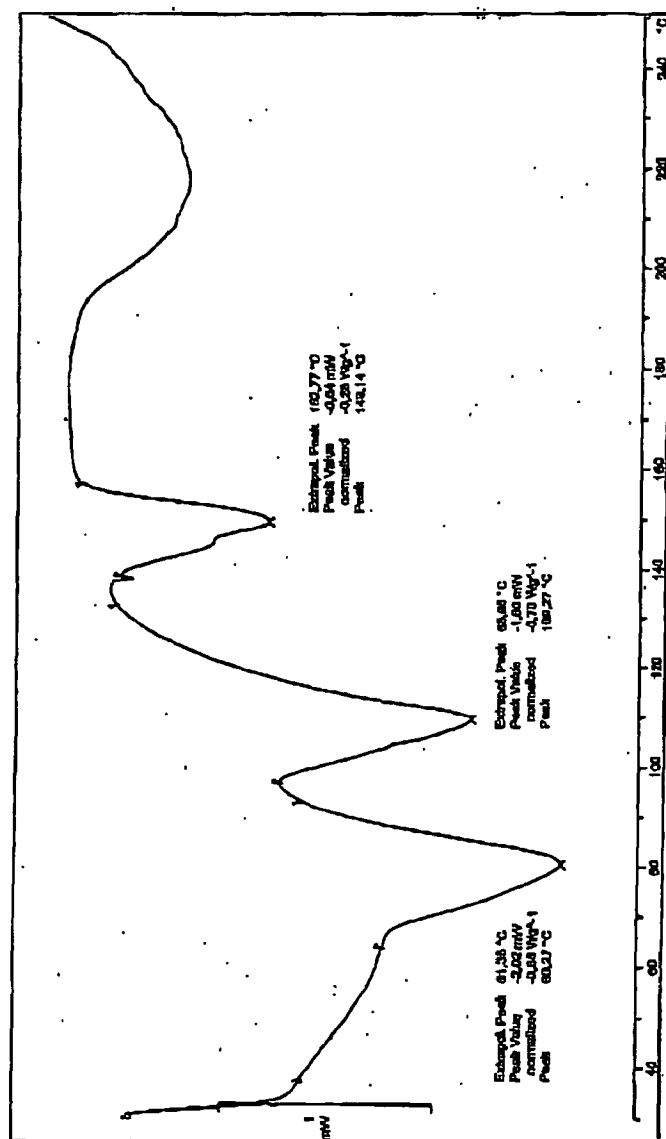


FIGURE 2

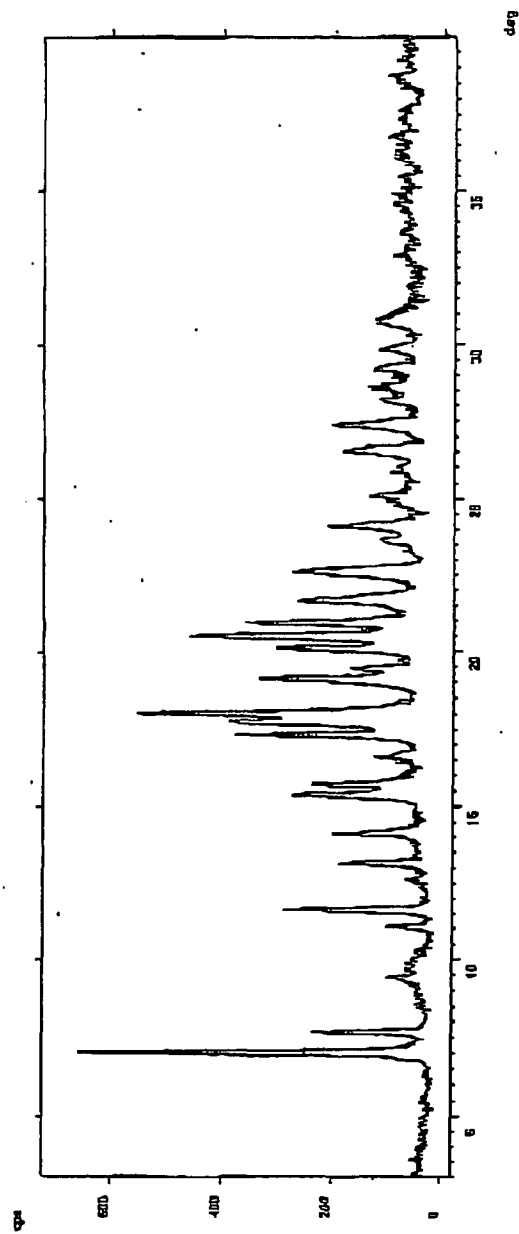


FIGURE 3

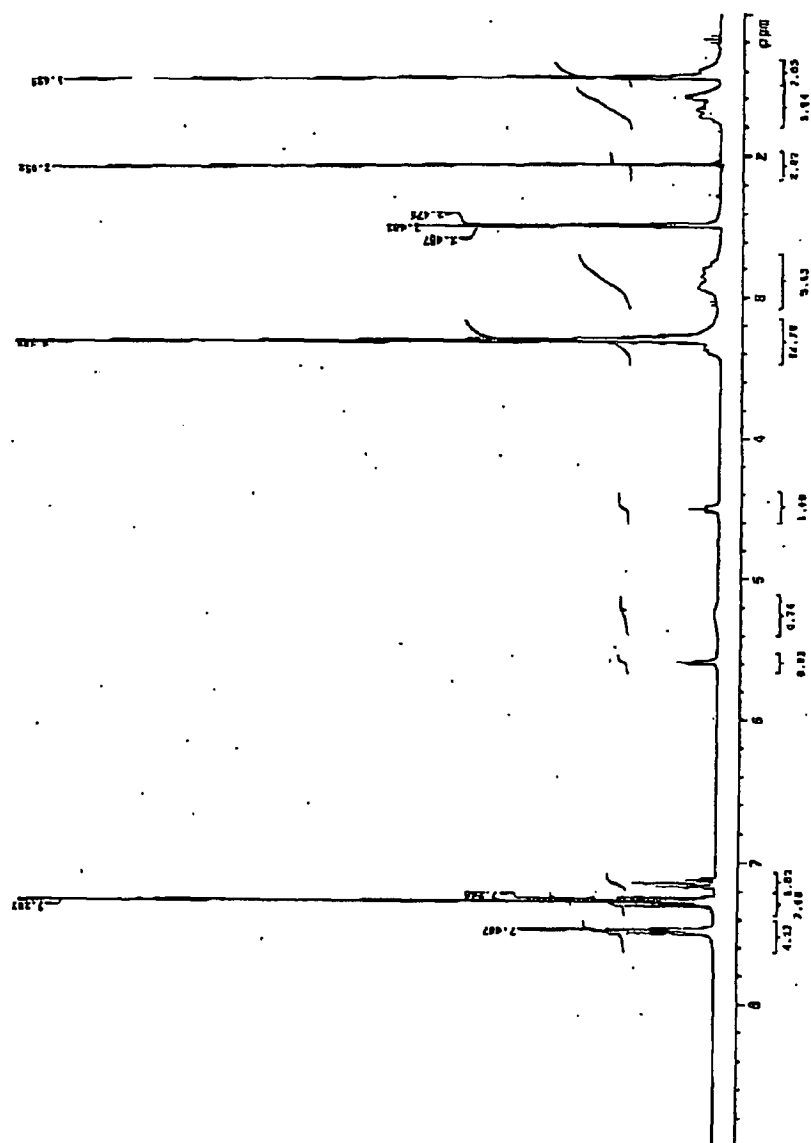


FIGURE 4



European Patent
Office

EUROPEAN SEARCH REPORT

Application Number
EP 05 01 1652

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
D,Y	WO 02/064429 A (CILAG AG) 29 August 2002 (2002-08-29) * page 2, line 4 - line 33; claims; examples *	1-16	C07D211/22 A61K31/4409 A61P11/08
D,Y	US 2003/021849 A1 (DOLITZKY ET. AL.) 30 January 2003 (2003-01-30) * claims; examples *	1-16	
Y	WO 02/102777 A (DR. REDDY'S LABORATORIES LTD.) 27 December 2002 (2002-12-27) * claims; examples *	1-16	
Y	WO 03/104197 A (TEVA PHARMACEUTICAL INDUSTRIES LTD.) 18 December 2003 (2003-12-18) * claims; examples *	1-16	
Y	WO 03/011295 A (TEX-CONTOR ETABLISSEMENT) 13 February 2003 (2003-02-13) * claims; examples *	1-16	
D,Y	WO 95/31437 A (MARION MERRELL DOW INC.) 23 November 1995 (1995-11-23) * claims; examples *	1-16	C07D A61K A61P
P,Y	WO 2005/019175 A (CIPLA LTD.) 3 March 2005 (2005-03-03) * claims; examples *	1-16	
The present search report has been drawn up for all claims			
Place of search Munich		Date of completion of the search 5 October 2005	Examiner Helps, I
<p>CATEGORY OF CITED DOCUMENTS</p> <p>X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document</p> <p>T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons</p> <p>& : member of the same patent family, corresponding document</p>			

1
EPO FORM : 503 03 02 (P04G01)

**ANNEX TO THE EUROPEAN SEARCH REPORT
ON EUROPEAN PATENT APPLICATION NO.**

EP 05 01 1652

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report.
The members are as contained in the European Patent Office EDP file on
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

05-10-2005

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
WO 02064429	A	22-08-2002	AT 300474 T	15-08-2005
			DE 60112329 D1	01-09-2005
			EP 1358109 A1	05-11-2003
			IT MI20010270 A1	09-08-2002
			US 2005011166 A1	20-01-2005

US 2003021849	A1	30-01-2003	US 2002177608 A1	28-11-2002
			US 2004058955 A1	25-03-2004

WO 02102777	A	27-12-2002	BG 108435 A	30-12-2004
			BR 0117054 A	27-07-2004
			CA 2450858 A1	27-12-2002
			CN 1518540 A	04-08-2004
			CZ 20033358 A3	14-04-2004
			EE 200400010 A	16-02-2004
			EP 1399422 A2	24-03-2004
			HU 0401546 A2	28-12-2004
			JP 2005507374 T	17-03-2005
			SK 15372003 A3	03-08-2004
			ZA 200309557 A	14-09-2004

WO 2003104197	A	18-12-2003	AU 2003248657 A1	22-12-2003
			DE 03757471 T1	01-09-2005
			EP 1474392 A1	10-11-2004

WO 2003011295	A	13-02-2003	US 2004248935 A1	09-12-2004

WO 9531437	A	23-11-1995	AT 220667 T	15-08-2002
			AU 693892 B2	09-07-1998
			AU 2426595 A	05-12-1995
			CA 2189007 A1	23-11-1995
			CN 1148849 A	30-04-1997
			CN 1623985 A	08-06-2005
			DE 69527429 D1	22-08-2002
			DE 69527429 T2	20-03-2003
			DK 766668 T3	28-10-2002
			EP 0766668 A1	09-04-1997
			ES 2176329 T3	01-12-2002
			FI 964565 A	14-11-1996
			HU 76134 A2	30-06-1997
			IL 113747 A	20-05-2001
			JP 10500134 T	06-01-1998
			JP 2002308849 A	23-10-2002
			JP 2002255935 A	11-09-2002
			JP 2002308850 A	23-10-2002
			JP 2002255936 A	11-09-2002

EPO FORM P0459

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82

ANNEX TO THE EUROPEAN SEARCH REPORT
ON EUROPEAN PATENT APPLICATION NO.

EP 05 01 1652

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report.
The members are as contained in the European Patent Office EDP file on
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

05-10-2005

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
WO 9531437	A	JP 2002316978 A	31-10-2002
		NO 964859 A	15-11-1996
		PT 766668 T	29-11-2002

WO 2005019175	A	03-03-2005	NONE

EPO FORM P449

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82


```
#####
#      #
#      #
#      #
#      #
#      #
#      #
#####
##  ##  ##  ##  ##  ##
##  #   ##  #   ##
##  #   ##  #   ##
##  #   ##  #   ##
##  #   ##  #   ##
##  #   ##  #   ##
#####
```

```
#####
#      #
#      #
#      #
#      #
#      #
#      #
#####
##  ##  ##  ##  ##  ##
##  #   ##  #   ##
##  #   ##  #   ##
##  #   ##  #   ##
##  #   ##  #   ##
##  #   ##  #   ##
#####
```


Print Job Information:

Date: 4/26/2007

Time: 4:43:33 PM

Job Number: 138

Reminder: AOL will never ask you to send us your password or credit card number in an email. This message has been scanned for known viruses.

From: johnconnor6@aol.com
To: editor@trainsmag.com
Cc: Johnconnor6@aol.com
Subject: CORRECTION
Date: Thu, 5 Apr 2007 2:53 PM

To the Editor,

On page 13 of May issue, reference is made to the 1939 Royal train tour: "in 1939...Canadian Pacific and Canadian National partnered to take King George IV and Queen Elizabeth across Canada."

I believe the king was King George **VI**.

John Connor

AOL now offers free email to everyone. Find out more about what's free from AOL at [AOL.com](http://aol.com).